

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
17 January 2002 (17.01.2002)

PCT

(10) International Publication Number
WO 02/04119 A1

(51) International Patent Classification⁷: **B01J 31/18**,
31/34, 31/24, C07F 9/50, 9/46, C07C 2/36, 11/107, C08F
10/02

MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK,
SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA,
ZW.

(21) International Application Number: PCT/GB01/03006

(84) Designated States (regional): ARIPO patent (GH, GM,
KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian
patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European
patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE,
IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF,
CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

(22) International Filing Date: 4 July 2001 (04.07.2001)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
0016895.5 11 July 2000 (11.07.2000) GB

(71) Applicant (for all designated States except US): **BP CHEMICALS LIMITED [GB/GB]**; Britannic House, 1 Finsbury Circus, London EC2M 7BA (GB).

(72) Inventor; and

(75) Inventor/Applicant (for US only): **WASS, Duncan, Frank [GB/GB]**; 42 Sherland Road, Twickenham, Middlesex, TW1 4HD (GB).

(74) Agent: **SMITH, Julian, Philip, Howard**; BP International Limited, Chertsey Road, Sunbury-on-Thames, Middlesex TW16 7LN (GB).

(81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG)

— of inventorship (Rule 4.17(iv)) for US only

Published:

— with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

A1

WO 02/04119

(54) Title: OLEFIN TRIMERISATION USING A CATALYST COMPRISING A SOURCE OF CHROMIUM, MOLYBDENUM OR TUNGSTEN AND A LIGAND CONTAINING AT LEAST ONE PHOSPHOROUS, ARSENIC OR ANTIMONY ATOM BOUND TO AT LEAST ONE (HETERO)HYDROCARBYL GROUP

(57) Abstract: A process for the trimerisation of olefins is disclosed, comprising contacting a monomeric olefin or mixture of olefins under trimerisation conditions with a catalyst which comprises (a) a source of chromium, molybdenum or tungsten (b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but excluding the case where all such polar substituents are phosphane, arsane or stibane groups; and optionally (c) an activator.

OLEFIN TRIMERISATION USING A CATALYST COMPRISING A SOURCE OF CHROMIUM,
MOLYBDENUM OR TUNGSTEN AND A LIGAND CONTAINING AT LEAST ONE PHOSPHOROUS,
ARSENIC OR ANTIMONY ATOM BOUND TO AT LEAST ONE (HETERO)HYDROCARBYL GROUP

This invention relates to the trimerisation of olefins, such as the preparation of 1-hexene by the trimerisation of ethylene.

US 5198563 and related patents by Phillips describe chromium-containing catalysts containing monodentate amide ligands useful for trimerising olefins.

5 US 5968866 discloses an ethylene oligomerisation/trimerisation process which uses a catalyst comprising a chromium complex which contains a coordinating asymmetric tridentate phosphane, arsane or stibane ligand (referred to therein as phosphine, arsine or stibine, and representing a phosphorus, arsenic or antimony atom attached to three hydrocarbyl groups) and an aluminoxane to produce alpha-olefins
10 which are enriched in 1-hexene. There is no suggestion that it is possible to replace any of the phosphane, arsane or stibane groups: indeed, it is impossible to predict what the effect of such a replacement would be.

We have now discovered further ligands which when used in conjunction with a source of a Group 3 to 10 transition metal are significantly more active as trimerisation
15 catalysts than those currently known, and also show other advantageous properties. The invention also encompasses within its scope novel catalysts comprising such ligands in conjunction with a source of chromium, molybdenum or tungsten.

Accordingly in a first aspect, the present invention provides a catalyst for the trimerisation of olefins, comprising

20 (a) a source of chromium, molybdenum or tungsten;
(b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but

excluding the case where all such polar substituents are phosphane, arsane or stibane groups; and optionally

- (c) an activator.

In this specification the term "trimerisation" means catalytic reaction of a single olefinic monomer or a mixture of olefinic monomers to give products enriched in those constituents derived from the reaction(s) of three olefinic monomers, as distinct from polymerisation or oligomerisation, which typically give olefinic product distributions governed by either a geometric series equation or following a Poisson pattern of distribution. "Trimerisation" includes the case where all the monomer units in the trimerisation product are identical, where the trimerization product is made from two different olefins (i.e. two equivalents of one monomer react with one equivalent of a second monomer), and also where three different monomer units react to yield the product. A reaction involving more than one monomer is often referred to as cotrimerisation.

It will be appreciated that the above catalyst may either be formed prior to use in a trimerisation reaction, or it may be formed in situ by adding the individual components thereof to the reaction mixture.

In a further aspect, the invention provides a process for the trimerisation of olefins, comprising contacting a monomeric olefin or mixture of olefins under trimerisation conditions with a catalyst which comprises

- (a) a source of a Group 3 to 10 transition metal;
- (b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but excluding the case where all such polar substituents are phosphane, arsane or stibane groups; and optionally
- (c) an activator.

We have also found that the catalysts used in the above process have certain novel features. For example, such catalysts when supported lose less of their activity compared with the equivalent unsupported catalyst than known catalysts. A further aspect of the invention therefore is a supported catalyst having a productivity per mole of catalyst of at least 50%, preferably at least 70% of its productivity when unsupported, which catalyst preferably comprises

- (a) a source of a Group 3 to 10 transition metal;
- (b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but excluding the case where all such polar substituents are phosphane, arsane or stibane

5 groups; and optionally

- (c) an activator.

Additionally, we have found that such catalysts have unusually high productivity, and maintain that productivity particularly well. Accordingly one further aspect of the invention comprises a catalyst for the trimerisation of olefins, which has a productivity 10 of at least 30000g product per mmol catalyst per hour at a temperature of 110°C or less and an ethylene partial pressure of 21 bar or less. Another aspect of the invention is a catalyst for the trimerisation of olefins, wherein the catalyst productivity decays at a rate of less than 10% per hour.

In one embodiment of the process of the invention, the catalyst utilised in the 15 present invention additionally comprises a further catalyst (d) suitable for the polymerisation, oligomerisation or other chemical transformations of olefins. In processes wherein such an additional catalyst is present, the trimerisation products are incorporated into a higher polymer or other chemical product.

The catalysts used in the trimerisation process of the invention show 20 exceptionally high productivity and selectivity to 1-hexene within the product fraction containing 6 carbon atoms. The high productivity of the catalysts results in greater process efficiency and/or lower intrinsic levels of catalyst residues. The high selectivity of the catalysts results in a greater ease of product purification (resulting either in less costly product purification or purer products). These advantages would be expected to 25 apply both to processes wherein catalysts according to the invention comprise the sole catalytic component and also to integrated processes, for example in the production of branched polyolefins, where more than one transition metal catalyst is employed.

As regards the source of Group 3 to 10 transition metal (a), this can include simple 30 inorganic and organic salts, for example, halides, acetylacetones, carboxylates, oxides, nitrates, sulfates and the like, as well as co-ordination and organometallic complexes, for example, chromium trichloride tetrahydrofuran complex, (benzene)tricarbonylchromium, chromium hexacarbonyl, molybdenum hexacarbonyl

and the like. Preferably component (a) is a source of chromium, molybdenum or tungsten; particularly preferred is chromium.

The ligand of component (b) preferably has the formula

$(R^1)(R^2)X - Y - X(R^3)(R^4)$, wherein

5 X is phosphorus, arsenic or antimony;

Y is a linking group;

and R^1 , R^2 , R^3 and R^4 are each independently hydrocarbyl, substituted hydrocarbyl, heterohydrocarbyl or substituted heterohydrocarbyl groups, at least one of which has a polar substituent which is not a phosphane, arsane or stibane group.

10 An alternative preferred structure for the ligand of component (b) is $X(R^1)(R^2)(R^3)$ wherein X and R^1 , R^2 and R^3 are as defined above, with at least one of R^1 , R^2 and R^3 having a polar substituent which is not a phosphane, arsane or stibane group.

X is preferably phosphorus. As regards R^1 , R^2 , R^3 and R^4 , examples of suitable hydrocarbyl groups are methyl, ethyl, ethylenyl, propyl, butyl, cyclohexyl, benzyl, phenyl, tolyl, xylyl, mesityl, biphenyl, naphthyl, anthracenyl and the like. Examples of suitable heterohydrocarbyl groups are methoxy, ethoxy, phenoxy (i.e. $-OC_6H_5$), tolyloxy (i.e. $-OC_6H_4(CH_3)$), xylyloxy, mesityloxy, dimethylamino, diethylamino, methylmethylethylamino, thiomethyl, thiophenyl, trimethylsilyl, dimethylhydrazyl and the like.

Preferably those of R^1 to R^4 having polar substituents are substituted aryl groups with at least one polar substituent. Suitable substituted aryl groups include substituted phenyl, substituted naphthyl and substituted anthracenyl groups. Substituted phenyl is preferred. Polar substituents include methoxy, ethoxy, isopropoxy, C_3-C_{20} alkoxy, phenoxy, pentafluorophenoxy, trimethylsiloxy, dimethylamino, methylsulphanyl, tosyl, methoxymethyl, methylthiomethyl, 1,3-oxazolyl, methoxymethoxy, hydroxyl, amino, sulphate, nitro and the like. Other suitable polar substituents include phosphanes, arsanes and stibanes as described in US 5968866 (but subject to the above-mentioned proviso that at least one of R^1 to R^4 has a polar substituent which is not one of these). Ortho-substituted phenyl groups are most preferred; the ortho substituent is preferably alkoxy, more preferably methoxy or methoxymethoxy. The phenyl groups may additionally be substituted in the meta and para or other ortho positions by groups such as hydrocarbyl, heterohydrocarbyl, substituted hydrocarbyl, halide and the like; but it is preferred that they are unsubstituted in these other positions.

Preferably any of R¹ to R⁴ which do not have polar substituents are independently optionally substituted phenyl groups; substituents may be hydrocarbyl, heterohydrocarbyl, substituted hydrocarbyl, substituted heterohydrocarbyl, halide and the like. However it is most preferred that all of R¹ to R⁴ have polar substituents as defined above, which are not phosphane, arsane or stibane groups. It is also most preferred that R¹ to R⁴ are the same.

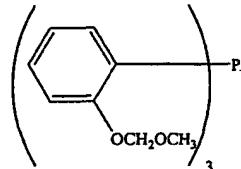
Y may be any bridging group, for example hydrocarbyl, substituted hydrocarbyl, heterohydrocarbyl, substituted hydrocarbyl or substituted heterohydrocarbyl bridging groups, or inorganic bridging groups including single atom links such as -O-. Y may 10 optionally contain an additional potential donor site. Examples of Y include methylene, 1,2-ethane, 1,2-phenylene, 1,3-propane, 1,2-catechol, 1,2-dimethylhydrazine, -N(R⁵)- where R⁵ is hydrogen, hydrocarbyl, or substituted hydrocarbyl, and the like. Preferably Y is -N(R⁵)-; preferably R⁵ is hydrogen, C₁-C₆ alkyl or phenyl, more preferably methyl.

Any of the groups R¹-R⁴ may independently be linked to one or more of each 15 other or to the bridging group Y, to form a cyclic structure together with X or X and Y.

The ligands can be prepared using procedures known to one skilled in the art and disclosed in published literature. Examples of preferred compounds are:

- (2-methoxyphenyl)(phenyl)PN(Me)P(phenyl)₂
- (2-methoxyphenyl)₂PN(Me)P(phenyl)₂
- 20 (2-methoxyphenyl)(phenyl)PN(Me)P(2-methoxyphenyl)(phenyl)
- (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂
- (2-ethoxyphenyl)₂PN(Me)P(2-ethoxyphenyl)₂
- (2-isopropoxyphenyl)₂PN(Me)P(2-isopropoxyphenyl)₂
- (2-hydroxyphenyl)₂PN(Me)P(2-hydroxyphenyl)₂
- 25 (2-nitrophenyl)₂PN(Me)P(2-nitrophenyl)₂
- (2,3-dimethoxyphenyl)₂PN(Me)P(2,3-dimethoxyphenyl)₂
- (2,4-dimethoxyphenyl)₂PN(Me)P(2,4-dimethoxyphenyl)₂
- (2,6-dimethoxyphenyl)₂PN(Me)P(2,6-dimethoxyphenyl)₂
- (2,4,6-trimethoxyphenyl)₂PN(Me)P(2,4,6-trimethoxyphenyl)₂
- 30 (2-dimethoxyphenyl)(2-methylphenyl)PN(Me)P(2-methylphenyl)₂
- [2-(dimethylamino)phenyl]₂PN(Me)P[2-(dimethylamino)phenyl]₂
- (2-methoxymethoxyphenyl)₂PN(Me)P(2-methoxymethoxyphenyl)₂

(2-methoxyphenyl)₂PN(Ethyl)P(2-methoxyphenyl)₂
 (2-methoxyphenyl)₂PN(Phenyl)P(2-methoxyphenyl)₂
 (2-methoxyphenyl)₂PN(Me)N(Me)P(2-methoxyphenyl)₂
 (2-methoxyphenyl)₂PCH₂P(2-methoxyphenyl)₂
 5 (2-methoxyphenyl)₂PCH₂CH₂P(2-methoxyphenyl)₂
 tri(2-methoxymethoxyphenyl)phosphane i.e.



tri(2-methoxyphenyl) phosphane.

Components (a) and (b) may be present in any ratio, preferably between 10000:1
 10 and 1:10000; more preferred is a ratio between 100:1 and 1:100, and especially preferred is a ratio of 10:1 to 1:10, particularly 3:1 to 1:3. Generally the amounts of (a) and (b) are approximately equal, ie a ratio of between 1.5:1 and 1:1.5.

The activator compound (c) may in principle be any compound that generates an active catalyst with components a) and b). Mixtures of activators may also be used.
 15 Suitable compounds include organoaluminium compounds, organoboron compounds and inorganic acids and salts, such as tetrafluoroboric acid etherate, silver tetrafluoroborate, sodium hexafluoroantimonate and the like. Suitable organoaluminium compounds include compounds of the formula AlR₃, where each R is independently C₁-C₁₂ alkyl, oxygen or halide, and compounds such as LiAlH₄ and the like. Examples
 20 include trimethylaluminium (TMA), triethylaluminium (TEA), tri-isobutylaluminium (TIBA), tri-n-octylaluminium, methylaluminium dichloride, ethylaluminium dichloride, dimethylaluminium chloride, diethylaluminium chloride, ethylaluminiumsesquichloride, methylaluminiumsesquichloride, and alumoxanes. Alumoxanes are well known in the art as typically oligomeric compounds which can be prepared by the controlled addition
 25 of water to an alkylaluminium compound, for example trimethylaluminium. Such compounds can be linear, cyclic, cages or mixtures thereof. Commercially available alumoxanes are generally believed to be mixtures of linear and cyclic compounds. The cyclic alumoxanes can be represented by the formula [R⁶AlO]_s and the linear alumoxanes by the formula R⁷(R⁸AlO)_s wherein s is a number from about 2 to 50, and
 30 wherein R⁶, R⁷, and R⁸ represent hydrocarbyl groups, preferably C₁ to C₆ alkyl groups,

for example methyl, ethyl or butyl groups. Alkylalumoxanes such as methylalumoxane (MAO) are preferred.

Mixtures of alkylalumoxanes and trialkylaluminium compounds are particularly preferred, such as MAO with TMA or TIBA. In this context it should be noted that the 5 term "alkylalumoxane" as used in this specification includes alkylalumoxanes available commercially which may contain a proportion, typically about 10wt%, but optionally up to 50wt%, of the corresponding trialkylaluminium; for instance, commercial MAO usually contains approximately 10wt% trimethylaluminium (TMA), whilst commercial MMAO contains both TMA and TIBA. Quantities of alkylalumoxane quoted herein 10 include such trialkylaluminium impurities, and accordingly quantities of trialkylaluminium compounds quoted herein are considered to comprise compounds of the formula AlR_3 additional to any AlR_3 compound incorporated within the alkylalumoxane when present.

Examples of suitable organoboron compounds are boroxines, NaBH_4 , 15 trimethylboron, triethylboron, dimethylphenylammoniumtetra(phenyl)borate, trityltetra(phenyl)borate, triphenylboron, dimethylphenylammonium tetra(pentafluorophenyl)borate, sodium tetrakis[(bis-3,5-trifluoromethyl)phenyl]borate, $\text{H}^+(\text{OEt}_2)_2$ [(bis-3,5-trifluoromethyl)phenyl]borate, trityltetra(pentafluorophenyl)borate and tris(pentafluorophenyl) boron.

20 Activator compound (c) may also be or contain a compound that acts as a reducing or oxidising agent, such as sodium or zinc metal and the like, or oxygen and the like.

In the preparation of the catalysts utilised in the present invention, the quantity of activating compound to be employed is easily determined by simple testing, for 25 example, by the preparation of small test samples which can be used to trimerise small quantities of the monomer(s) and thus to determine the activity of the produced catalyst. It is generally found that the quantity employed is sufficient to provide 0.1 to 20,000 atoms, preferably 1 to 2000 atoms of aluminium or boron per atom of chromium. In some cases, for particular combinations of component a) and b), an activating 30 compound c) may not be required.

Components (a)-(c) of the catalyst system utilised in the present invention may be added together simultaneously or sequentially, in any order, and in the presence or

absence of monomer in any suitable solvent, so as to give an active catalyst. For example, components (a), (b) and (c) and monomer may be contacted together simultaneously, or components (a), (b) and (c) may be added together simultaneously or sequentially in any order and then contacted with monomer, or components a) and b)

5 may be added together to form an isolable metal-ligand complex and then added to component c) and contacted with monomer, or components (a), (b) and (c) may be added together to form an isolable metal-ligand complex and then contacted with monomer. Suitable solvents for contacting the components of the catalyst or catalyst system include, but are not limited to, hydrocarbon solvents such as heptane, toluene , 1-
10 hexene and the like, and polar solvents such as diethyl ether, tetrahydrofuran, acetonitrile, dichloromethane, chloroform, chlorobenzene, methanol, acetone and the like.

The catalyst components (a), (b) and (c) utilised in the present invention can be unsupported or supported on a support material, for example, silica, alumina, MgCl₂ or
15 zirconia, or on a polymer, for example polyethylene, polypropylene, polystyrene, or poly(aminostyrene). It is an advantage of the present invention that very little productivity (mass of product per mol of catalyst per hour) is lost when the catalyst is supported. If desired the catalysts can be formed in situ in the presence of the support material, or the support material can be pre-impregnated or premixed, simultaneously or
20 sequentially, with one or more of the catalyst components. The quantity of support material employed can vary widely, for example from 100,000 to 1 grams per gram of metal present in the transition metal compound. In some cases, the support may material can also act as or as a component of the activator compound (c). Examples include supports containing alumoxane moieties and/or hydrocarbyl boryl moieties (see, for
25 example, Hlatky, G.G. *Chem. Rev.* 2000, 100, 1347.)

One embodiment of the present invention encompasses the use of components (a) (b) and optionally (c) in conjunction with one or more types of olefin polymerisation catalyst or catalyst system (d) to trimerise olefins and subsequently incorporate a portion of the trimerisation product(s) into a higher polymer.

30 Component (d) may be one or more suitable polymerisation catalyst(s) or catalyst system(s), examples of which include, but are not limited to, conventional Ziegler-Natta catalysts, metallocene catalysts, monocyclopentadienyl or "constrained geometry"

catalysts, heat activated supported chromium oxide catalysts (eg. "Phillips"-type catalysts), late transition metal polymerisation catalysts (eg. diimine, diphosphine and salicylaldimine nickel/palladium catalysts, iron and cobalt pyridyldiimine catalysts and the like) and other so-called "single site catalysts" (SSC's).

5 Ziegler-Natta catalysts, in general, consist of two main components. One component is an alkyl or hydride of a Group I to III metal, most commonly Al(Et)₃ or Al(iBu)₃ or Al(Et)₂Cl but also encompassing Grignard reagents, n-butyllithium, or dialkylzinc compounds. The second component is a salt of a Group IV to VIII transition metal, most commonly halides of titanium or vanadium such as TiCl₄, TiCl₃, VCl₄, or 10 VOCl₃. The catalyst components when mixed, usually in a hydrocarbon solvent, may form a homogeneous or heterogeneous product. Such catalysts may be impregnated on a support, if desired, by means known to those skilled in the art and so used in any of the major processes known for co-ordination catalysis of polyolefins such as solution, slurry, and gas-phase. In addition to the two major components described above, 15 amounts of other compounds (typically electron donors) may be added to further modify the polymerization behaviour or activity of the catalyst.

Metallocene catalysts, in general, consist of transition metal complexes, most commonly based on Group IV metals, ligated with cyclopentadienyl(Cp)-type groups. A wide range of structures of this type of catalysts is known, including those with 20 substituted, linked and/or heteroatom-containing Cp groups, Cp groups fused to other ring systems and the like. Additional activators, such as boranes or alumoxane, are often used and the catalysts may be supported, if desired.

Monocyclopentadienyl or "constrained geometry" catalysts, in general, consist of a transition metal complexes, most commonly based on Group IV metals, ligated 25 with one cyclopentadienyl(Cp)-type group, often linked to additional donor group. A wide range of structures of this type of catalyst is known, including those with substituted, linked and/or heteroatom-containing Cp groups, Cp groups fused to other ring systems and a range of linked and non-linked additional donor groups such as amides, amines and alkoxides. Additional activators, such as boranes or alumoxane, are 30 often used and the catalysts may be supported, if desired.

A typical heat activated chromium oxide (Phillips) type catalyst employs a combination of a support material to which has first been added a chromium-containing

material wherein at least part of the chromium is in the hexavalent state by heating in the presence of molecular oxygen. The support is generally composed of about 80 to 100 wt.% silica, the remainder, if any, being selected from the group consisting of refractory metal oxides, such as aluminium, boria, magnesia, thoria, zirconia, titania and mixtures of two or more of these refractory metal oxides. Supports can also comprise alumina, aluminium phosphate, boron phosphate and mixtures thereof with each other or with silica. The chromium compound is typically added to the support as a chromium (III) compound such as the acetate or acetylacetone in order to avoid the toxicity of chromium (VI). The raw catalyst is then calcined in air at a temperature between 250 and 1000°C for a period of from a few seconds to several hours. This converts at least part of the chromium to the hexavalent state. Reduction of the Cr (VI) to its active form normally occurs in the polymerization reaction, but can be done at the end of the calcination cycle with CO at about 350°C. Additional compounds, such as fluorine, aluminium and/or titanium may be added to the raw Phillips catalyst to modify it.

15 Late transition metal and single site catalysts cover a wide range of catalyst structures based on metals across the transition series (see, for example, Britovsek, G.J.P et al. *Angew. Chem. Int. Ed. Engl.* 1999, 38, 429. and Ittel, S.D. et al. *Chem. Rev.* 2000, 100, 1169.

20 Component (d) may also comprise one or more polymerisation catalysts or catalyst systems together with one or more additional oligomerisation catalysts or catalyst systems. Suitable oligomerisation catalysts include, but are not limited to, those that dimerise (for example, nickel phosphine dimerisation catalysts) or trimerise olefins or otherwise oligomerise olefins to, for example, a distribution of 1-olefins governed by a geometric series equation (for example, iron and cobalt pyridylidinium oligomerisation catalysts).

25 Component (d) may independently be supported or unsupported. Where components (a) and (b) and optionally (c) are supported, (d) may be co-supported sequentially in any order or simultaneously on the same support or may be on a separate support. For some combinations, the components (a) - (c) may be part or all of component (d). For example, if component (d) is a heat activated chromium oxide catalyst then this may be (a), a chromium source and if component (d) contains an alumoxane activator then this may also be the optional activator (c).

The components (a), (b), (c) and (d) may be in any molar ratio. In the context of an integrated process the ratio of (a) to (d) is seen as particularly important. The ratio of (a) to (d) is preferably from 10000:1 to 1:10000 and more preferably from 100:1 to 1:100. The precise ratio required depends on the relative reactivity of the components and also 5 on the desired properties of the product or catalyst systems.

Suitable olefinic monomers, or combinations thereof for use in the trimerisation process of the present invention are hydrocarbon olefins, for example, ethylene, C₂₋₂₀ α-olefins, internal olefins, vinylidene olefins, cyclic olefins and dienes, propylene, 1-butene, 1-pentene, 1-hexene, 4-methylpentene-1, 1-heptene, 1-octene, 1-nonene, 1-decene, 1-undecene, 1-dodecene, 1-tridecene, 1-tetradecene, 1-pentadecene, 1-hexadecene, 1-heptadecene, 1-octadecene, 1-nonadecene, 1-eicosene, styrene, 2-butene, 2-ethyl-1-hexene, cyclohexene, norbornene, butadiene and 1,5-hexadiene. Olefins with a polar functionality, such as methyl (meth)acrylate, vinyl acetate, α,ω-undecenol and the like, may also be used. The preferred monomer is ethylene. Mixtures of these 10 monomer may also be used, for example a 1-butene unit and two ethylene units may be co-trimerised to form C8 olefins, or 1-hexene and ethylene co-trimerised to C10 olefins, or 1-dodecene and ethylene co-trimerised to C16 olefins. Combinations of these co-trimerisation reactions may be performed simultaneously, especially when one or more of the monomers are produced in-situ (e.g. a mixture of ethylene and butene can 15 be used to form mixtures containing predominantly hexenes, octenes, and decenes.) Techniques for varying the distribution of products from these reactions include controlling process conditions (e.g. concentration, reaction temperature, pressure, residence time) and properly selecting the design of the process and are well known to those skilled in the art. These monomers or combinations thereof are also suitable in the 20 presence of component (d).

Olefinic monomers or mixtures of olefinic monomers for trimerisation may be substantially pure or may contain olefinic impurities. One embodiment of the process of the invention comprises the trimerisation of olefin-containing waste streams from other chemical processes or other stages of the same process.

When operating under solution or slurry phase conditions, any diluent or solvent 30 that is an olefin, a mixture of olefins, or is substantially inert under trimerisation conditions may be employed. Mixtures of inert diluents, with or without one or more

olefins, also could be employed. The preferred diluents or solvents are aliphatic and aromatic hydrocarbons and halogenated hydrocarbons such as, for example, isobutane, pentane, toluene, xylene, ethylbenzene, cumene, mesitylene, heptane, cyclohexane, methylcyclohexane, 1-hexene, 1-octene, chlorobenzene, dichlorobenzene, and the like, and mixtures such as isopar.

The trimerisation conditions can be, for example, solution phase, slurry phase, gas phase or bulk phase, with temperatures ranging from -100°C to +300°C, preferably from 0°C to +300°C and more preferably from 35°C to 200°C, and at pressures of atmospheric and above, preferably from atmospheric to 800 barg and more preferably from 1 barg to 100 barg. If desired, the process can be operated at temperatures above 120°C, and optionally also at pressures below 30 barg. The high initial rate and low rate of deactivation of this catalyst system enables lower pressures to be employed than would have been economically feasible with prior art catalyst systems.

Irrespective of the trimerisation technique employed, trimerisation is typically carried out under conditions that substantially exclude oxygen, water, and other materials that act as catalyst poisons. Also, trimerisation can be carried out in the presence of additives to control selectivity, enhance activity and reduce the amount of polymer formed in trimerisation processes. Suitable additives include, but are not limited to, hydrogen or a halide source such as GeCl₄. Exemplary halides include, but are not limited to fluoride, chloride, bromide, and/or iodide.

There exist a number of options for the trimerisation reactor including batch, semi- batch, and continuous operation. The trimerisation and co-trimerisation reactions of the present invention can be performed under a range of process conditions that are readily apparent to those skilled in the art: as a homogeneous liquid phase reaction in the presence or absence of an inert hydrocarbon diluent such as toluene or heptanes; as a two-phase liquid/liquid reaction; as a slurry process where the catalyst is in a form that displays little or no solubility; as a bulk process in which essentially neat reactant and/or product olefins serve as the dominant medium; as a gas-phase process in which at least a portion of the reactant or product olefin(s) are transported to or from a supported form of the catalyst via the gaseous state. Evaporative cooling from one or more monomers or inert volatile liquids is but one method that can be employed to effect the removal of heat from the reaction. The trimerisation reactions may be performed in the known types

of gas-phase reactors, such as circulating bed, vertically or horizontally stirred-bed, fixed-bed, or fluidised-bed reactors, liquid-phase reactors, such as plug-flow, continuously stirred tank, or loop reactors, or combinations thereof. A wide range of methods for effecting product, reactant, and catalyst separation and/or purification are known to those skilled in the art and may be employed: distillation, filtration, liquid-liquid separation, slurry settling, extraction, etc. One or more of these methods may be performed separately from the trimerisation reaction or it may be advantageous to integrate at least some with a trimerisation reaction; a non-limiting example of this would be a process employing catalytic (or reactive) distillation. Also advantageous may be a process which includes more than one reactor, a catalyst kill system between reactors or after the final reactor, or an integrated reactor/separator/purifier. While all catalyst components, reactants, inert, and products could be employed in the present invention on a once-through basis, it is often economically advantageous to recycle one or more of these materials; in the case of the catalyst system, this might require reconstituting one or more of the catalysts components to achieve the active catalyst system. It is within the scope of this invention that a trimerisation product might also serve as a reactant (e.g. 1-hexene, produced via the trimerization of ethylene, might be converted to decene products via a subsequent co-trimerization reaction with ethylene.)

A number of process options can be envisaged when using the catalysts of the present invention in an integrated process that includes a subsequent chemical transformation, i.e. with component (d) present. These options include "in series" processes in which the trimerisation and subsequent reaction are performed in separate, linked reactors, optionally with recycling of products/reagents between the reactors, and "in situ" processes in which both reaction steps are carried out in the same reactor.

Chemical transformations involving olefins are well known to those skilled in the art: non-limiting examples of the chemical reactions that might be effected by use of a component (d) include polymerisation and co-polymerisation, oligomerisation, hydrogenation, hydroformylation, oxidation, hydration, sulfonation, epoxidation, isomerisation, amination, cyclisation, and alkylation. A typical reactor residence time in the polymerisation reactor is less than 4 hours, preferably less than 3 hours.

In the case of an "in series" process various purification, analysis and control steps for the oligomeric product could potentially be incorporated between the

trimerization and subsequent reaction stages. Recycling between reactors configured in series is also possible. An example of such a process would be the trimerisation of ethylene in a single reactor with a catalyst comprising components (a), (b) and optionally (c) followed by polymerisation of the trimerisation product with ethylene in a separate, linked reactor to give branched polyethylene. Another example would be co-trimerisation of ethylene and 1-butene and subsequent polymerisation of the trimerisation product to give poly(octene). Another example would be the trimerisation of an ethylene-containing waste stream from a polyethylene process, followed by introduction of the product 1-hexene back into the polyethylene process as a co-monomer for the production of branched polyethylene.

An example of an "in situ" process is the production of branched polyethylene catalysed by components (a), (b), (d) and optionally (c), added in any order such that the active catalytic species derived from components (a), (b) and optionally (c) is/are at some point present in a reactor with component (d)

Both the "in series and "in situ" approaches can be adaptions of current polymerisation technology for the process stages including component (d). All major olefin existing polymerisation processes, including multiple reactor processes, are considered adaptable to this approach. One adaption is the incorporation of a trimerisation catalyst bed into a recycle loop of a gas phase polymerisation process, this could be as a side or recycle stream within the main fluidisation recycle loop and or within the degassing recovery and recycle system.

Polymerisation conditions when component (d) is present can be, for example, solution phase, slurry phase, gas phase or bulk phase, with temperatures ranging from -100°C to +300°C, and at pressures of atmospheric and above, particularly from 1.40 to 41 bar. Reaction conditions, will typically have a significant impact upon the properties (e.g. density, melt index, yield) of the polymer being made and it is likely that the polymer requirements will dictate many of the reaction variables. Reaction temperature, particularly in processes where it is important to operate below the sintering temperature of the polymer, will typically, and preferably, be primarily selected to optimise the polymerisation reaction conditions. The high productivity, and kinetic profile characteristics, of this new trimerisation catalyst makes the 'in-situ' production of the comonomer, preferably hexene-1, during polymer, preferably polyethylene, production

far more commercially attractive than prior art catalysts systems. This is true even at the typical reaction temperatures and pressures for the production of polyethylenes with high comonomer contents such as LLDPE, VLDPE and ULDPE (preferably between 50°C and 100°C, depending upon the density of the polymer) and even when used in slurry and gas phase polymerisation processes (preferably total gas phase pressures between 15 and 30 bar and ethylene pressures between 10 and 70 percent of the gas phase). If desired, the catalyst can be used to polymerise ethylene under high pressure/high temperature process conditions wherein the polymeric material forms as a melt in supercritical ethylene. Preferably the polymerisation is conducted under gas phase fluidized bed or stirred bed conditions. Also, polymerisation or copolymerisation can be carried out in the presence of additives to control polymer or copolymer molecular weights. The use of hydrogen gas as a means of controlling the average molecular weight of the polymer or copolymer applies generally to the polymerization process of the present invention.

Slurry phase polymerisation conditions or gas phase polymerisation conditions are particularly useful for the production of high or low density grades of polyethylene, and polypropylene. In these processes the polymerisation conditions can be batch, continuous or semi-continuous. Furthermore, one or more reactors may be used, e.g. from two to five reactors in series. Different reaction conditions, such as different temperatures or hydrogen concentrations may be employed in the different reactors. In cascade operation the trimerisation catalyst may be added to any or all of the polymerisation reactors concerned. If added to the first reactor and carried through to subsequent reactors, the trimerisation catalyst may ,or may not, be supplemented in subsequent reactors with fresh trimerisation or polymerisation catalyst, it may be deactivated in subsequent reactors through addition of reversible or irreversible poisons that partially or fully kill the trimerisation catalyst or though addition of additional polymerisation catalysts or modifiers that deactivate the trimerisation catalyst.

In the slurry phase process and the gas phase process, the catalyst is generally supported and metered and transferred into the polymerization zone in the form of a particulate solid either as a dry powder (e.g. with an inert gas, ethylene or an olefin) or as a slurry. In addition, an optional activator can be fed to the polymerisation zone, for example as a solution, separately from or together with the solid catalyst. Components

(a)-(d) can be added to any part of the polymerisation reactor either on the same support particle or as a physical mixture on different support particles, or may be added separately to the same or different parts of the reactor sequentially in any order or simultaneously. Alternatively, (a)-(d) may be unsupported and independently added to

5 any part of the polymerisation reactor simultaneously or sequentially together or separately. The ratio of the primary monomer to the other (co)monomers has a significant impact on the properties of the polymer formed (eg density) and it is usually desirable to be tightly controlled. This ratio may be primarily controlled by altering the concentration or partial pressure of either the primary monomer and/or the

10 comonomer(s). Typically the primary monomer concentration will be controlled independently of the ratio to comonomers (for other reasons such as activity) and the primary monomer to comonomer ratio(s) may be controlled by varying the rate of introduction of trimerisation catalyst or by altering reaction conditions which preferentially impact the trimerisation reaction over the polymerisation reaction or

15 which impacts upon the distribution of comonomers actually formed (eg by using reversible poisons/activators). Fresh comonomer feed may additionally be introduced to the polymerisation reactor to control the ratio. It may be desirable to preferentially purge certain (co)monomer(s) that are formed in the trimerisation reaction by, for example, heating or cooling a vapour (or liquid) slip (or recycle) stream within the

20 polymerisation reaction (or degassing) systems. This may for example be optimised by controlling compressor knock-out or interstage conditions in recycle or degassing vent recovery compressors or by using dedicated condensing exchangers or distillation apparatus.

The rate of addition of each component may be independently controlled to

25 allow variations in the ratio of components and the density of the polymer produced. Pressure, temperature, hydrogen addition, halogenated hydrocarbon addition, electron donor addition, activator/retarder addition and other suitable variables may also be varied to control the activity of each component and also allow control of the polymer produced.

30 Once the polymer product is discharged from the reactor, any associated and absorbed hydrocarbons are substantially removed, or degassed, from the polymer by, for example, pressure let-down or gas purging using fresh or recycled steam, nitrogen or

light hydrocarbons (such as ethylene). Recovered gaseous or liquid hydrocarbons may be recycled to a purification system or the polymerisation zone.

In the slurry phase polymerisation process the polymerisation diluent is compatible with the polymer(s) and catalysts, and may be an alkane such as hexane, heptane, isobutane, or a mixture of hydrocarbons or paraffins. The polymerization zone can be, for example, an autoclave or similar reaction vessel, or a continuous liquid full loop reactor, e.g. of the type well-known in the manufacture of polyethylene by the Phillips Process. When the polymerisation process of the present invention is carried out under slurry conditions the polymerisation is preferably carried out at a temperature above 0°C, most preferably above 15°C. Under slurry conditions the polymerisation temperature is preferably maintained below the temperature at which the polymer commences to soften or sinter in the presence of the polymerisation diluent. If the temperature is allowed to go above the latter temperature, fouling of the reactor can occur. Adjustment of the polymerisation within these defined temperature ranges can provide a useful means of controlling the average molecular weight of the produced polymer. A further useful means of controlling the molecular weight is to conduct the polymerization in the presence of hydrogen gas which acts as chain transfer agent. Generally, the higher the concentration of hydrogen employed, the lower the average molecular weight of the produced polymer.

In bulk polymerisation processes, liquid monomer such as propylene is used as the polymerisation medium.

Methods for operating gas phase polymerisation processes are well known in the art. Such methods generally involve agitating (e.g. by stirring, vibrating or fluidising) a bed of catalyst, or a bed of the target polymer (i.e. polymer having the same or similar physical properties to that which it is desired to make in the polymerisation process) containing a catalyst, and feeding thereto a stream of monomer (under conditions such that at least part of the monomer polymerises in contact with the catalyst in the bed). The bed is generally cooled by the addition of cool gas (e.g. recycled gaseous monomer) and/or volatile liquid (e.g. a volatile inert hydrocarbon, or gaseous monomer which has been condensed to form a liquid). The polymer produced in, and isolated from, gas phase processes forms directly a solid in the polymerisation zone and is free from, or substantially free from liquid. As is well known to those skilled in the art, if any liquid

is allowed to enter the polymerisation zone of a gas phase polymerisation process the quantity of liquid in the polymerisation zone is small in relation to the quantity of polymer present.. This is in contrast to “solution phase” processes wherein the polymer is formed dissolved in a solvent, and “slurry phase” processes wherein the polymer forms as a suspension in a liquid diluent.

The gas phase process can be operated under batch, semi-batch, or so-called “continuous” conditions. It is preferred to operate under conditions such that monomer is continuously recycled to an agitated polymerisation zone containing polymerisation catalyst, make-up monomer being provided to replace polymerised monomer, and continuously or intermittently withdrawing produced polymer from the polymerisation zone at a rate comparable to the rate of formation of the polymer, fresh catalyst being added to the polymerisation zone to replace the catalyst withdrawn from the polymerisation zone with the produced polymer.

Methods for operating gas phase fluidized bed processes for making polyethylene, ethylene copolymers and polypropylene are well known in the art. The process can be operated, for example, in a vertical cylindrical reactor equipped with a perforated distribution plate to support the bed and to distribute the incoming fluidising gas stream through the bed. The fluidising gas circulating through the bed serves to remove the heat of polymerisation from the bed and to supply monomer for polymerization in the bed. Thus the fluidising gas generally comprises the monomer(s) normally together with some inert gas (e.g. nitrogen or inert hydrocarbons such as methane, ethane, propane, butane, pentane or hexane) and optionally with hydrogen as molecular weight modifier. The hot fluidising gas emerging from the top of the bed is led optionally through a velocity reduction zone (this can be a cylindrical portion of the reactor having a wider diameter) and, if desired, a cyclone and or filters to disentrain fine solid particles from the gas stream. The hot gas is then led to a heat exchanger to remove at least part of the heat of polymerisation. Catalysts are preferably fed continuously or at regular intervals to the bed. At start up of the process, the bed comprises fluidisable polymer which is preferably similar to the target polymer.

Polymer is produced continuously within the bed by the polymerization of the monomer(s). Preferably means are provided to discharge polymer from the bed continuously or at regular intervals to maintain the fluidized bed at the desired height.

The process is generally operated at relatively low pressure, for example, at 10 to 50 bars, and at temperatures for example, between 50 and 135°C. The temperature of the bed is maintained below the sintering temperature of the fluidized polymer to avoid problems of agglomeration.

5 In the gas phase fluidized bed process for polymerisation of olefins the heat evolved by the exothermic polymerisation reaction is normally removed from the polymerisation zone (i.e. the fluidised bed) by means of the fluidising gas stream as described above. The hot reactor gas emerging from the top of the bed is led through one or more heat exchangers wherein the gas is cooled. The cooled reactor gas, together
10 with any make-up gas, is then recycled to the base of the bed. In the gas phase fluidised bed polymerisation process of the present invention it is desirable to provide additional cooling of the bed (and thereby improve the space time yield of the process) by feeding a volatile liquid to the bed under conditions such that the liquid evaporates in the bed thereby absorbing additional heat of polymerisation from the bed by the "latent heat of
15 evaporation" effect. When the hot recycle gas from the bed enters the heat exchanger, the volatile liquid can condense out. In one embodiment of the present invention the volatile liquid is separated from the recycle gas and reintroduced separately into the bed. Thus, for example, the volatile liquid can be separated and sprayed into the bed. In another embodiment of the present invention the volatile liquid is recycled to the bed
20 with the recycle gas. Thus the volatile liquid can be condensed from the fluidising gas stream emerging from the reactor and can be recycled to the bed with recycle gas, or can be separated from the recycle gas and then returned to the bed.

The method of condensing liquid in the recycle gas stream and returning the mixture of gas and entrained liquid to the bed is described in EP-A-0089691 and EP-A-
25 0241947. It is preferred to reintroduce the condensed liquid into the bed separate from the recycle gas using the process described in our US 5541270.

A number of process options can be envisaged when using the catalysts of the present invention in an integrated process to prepare higher polymers i.e when component (d) is present. These options include "in series" processes in which the
30 trimerisation and subsequent polymerisation are carried in separate but linked reactors and "in situ" processes in which both reaction steps are carried out in the same reactor.

In the case of a gas phase "in situ" polymerisation process, component (d) can,

for example, be introduced into the polymerisation reaction zone in liquid form, for example, as a solution in a substantially inert liquid diluent. Components (a), (b), (c) and (d) may be independently added to any part of the polymerisation reactor simultaneously or sequentially together or separately. Under these circumstances it is 5 preferred the liquid containing the component(s) is sprayed as fine droplets into the polymerisation zone. The droplet diameter is preferably within the range 1 to 1000 microns. EP-A-0593083 discloses a process for introducing a polymerisation catalyst into a gas phase polymerization. The methods disclosed in EP-A-0593083 can be suitably employed in the polymerisation process of the present invention if desired.

10 Although not usually required, upon completion of polymerisation or copolymerisation, or when it is desired to terminate polymerisation or copolymerisation or at least temporarily deactivate the catalyst or catalyst component of this invention, the catalyst can be contacted with water, alcohols, acetone, or other suitable catalyst deactivators a manner known to persons of skill in the art.

15 The trimerisation catalyst is preferably (but optionally) added before the polymerization catalyst such that the desired primary monomer to comonomer(s) ratio is established prior to introduction of the polymerization catalyst. The desired comonomer composition at start-up may however be achieved through introduction of fresh comonomer feed or through judicious initiation of the trimerisation reaction before or 20 during polymerization catalyst introduction.

In the presence of component (d) the polymerisation process of the present invention provides polymers and copolymers, especially ethylene polymers, at high productivity (based on the amount of polymer or copolymer produced per unit weight of complex employed in the catalyst system). This means that relatively very small 25 quantities of transition metal complexes are consumed in commercial processes using the process of the present invention. It also means that when the polymerisation process of the present invention can be operated under polymer recovery conditions that do not employ a catalyst separation step, thus leaving the catalyst, or residues thereof, in the polymer (e.g. as occurs in most commercial slurry and gas phase polymerization 30 processes), the amount of transition metal complex in the produced polymer can be very small.

By varying the ratio of components (a) (b), optionally (c) and (d) and /or by

adding additional comonomers, catalysts of the present invention can provide a wide variety of branched polymers differing in density and in other important physical properties.

A range of polyethylene polymers are considered accessible including high density polyethylene, medium density polyethylene, low density polyethylene, ultra low density polyethylene and elastomeric materials. Particularly important are the polymers having a density in the range of 0.91 to 0.93, generally referred to in the art as linear low density polyethylene. Such polymers and copolymers are used extensively in the manufacture of flexible blown or cast film.

10 Poly(1-hexene), poly(1-octene) and the like are also considered accessible, as are copolymers of e.g. 1-hexene and propylene, 1-hexene and 1-octene and terpolymers of e.g. ethylene, 1-hexene and vinyl acetate.

Dienes could also be incorporated into the polymeric products to enable cross-linking for e.g. elastomer and wire and cable applications

15 Depending upon the use of the polymer product, minor amounts of additives are typically incorporated into the polymer formulation such as acid scavengers, antioxidants, stabilizers, and the like. Generally, these additives are incorporated at levels of about 25 to 2000 ppm, typically from about 50 to about 1000 ppm, and more typically 400 to 1000 ppm, based on the polymer.

20 In use, polymers or copolymers made according to the invention in the form of a powder are conventionally compounded into pellets. Examples of uses for polymer compositions made according to the invention include use to form fibres, extruded films, tapes, spunbonded webs, moulded or thermoformed products, and the like. The polymers may be blown or cast into films, or may be used for making a variety of moulded or extruded articles such as pipes, and containers such as bottles or drums. Specific additive packages for each application may be selected as known in the art. Examples of supplemental additives include slip agents, anti-blocks, anti-stats, mould release agents, primary and secondary anti-oxidants, clarifiers, nucleants, uv stabilizers, and the like. Classes of additives are well known in the art and include phosphite 25 antioxidants, hydroxylamine (such as N,N-dialkyl hydroxylamine) and amine oxide (such as dialkyl methyl amine oxide) antioxidants, hindered amine light (uv) stabilizers, phenolic stabilizers, benzofuranone stabilizers, and the like. Various olefin polymer

additives are described in U.S. patents 4,318,845, 4,325,863, 4,590,231, 4,668,721, 4,876,300, 5,175,312, 5,276,076, 5,326,802, 5,344,860, 5,596,033, and 5,625,090.

Fillers such as silica, glass fibers, talc, and the like, nucleating agents, and colourants also may be added to the polymer compositions as known by the art.

5 The present invention is illustrated in the following Examples.

EXAMPLES

All manipulations were performed under anaerobic conditions. Solvents and gases were dried and degassed by standard procedures. Chemicals were purchased from the Aldrich Chemical Company unless stated otherwise. Methyl alumoxane (MAO) and modified methyl alumoxane (MMAO) were purchased from Witco as 10% w/w solutions in toluene or heptanes respectively. (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ was synthesized by literature procedures (See example 12 of WO97/37765). Cr(p-tolyl)Cl₂(THF)₃ was synthesized by literature procedure (Daly, J. J.; Seeden, R.P.A.; J.Chem.Soc.A, 1967, 736). Reaction products were analysed by GCMS using 50m × 0.3mm id, CP sil. CBS-MS, df=0.4μm columns, an initial temperature of -30°C, hold 1 min, ramp rate 7 °C/min, final temperature 280°C and final hold of 5 mins. Molar quantities of catalyst are based upon the molar quantity of chromium source used in their preparation.

EXAMPLE 1

20 A Schlenk tube was charged with CrCl₃(THF)₃ (8 mg, 0.02 mmol) and (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ (10 mg, 0.02 mmol), 10ml THF was added and the solution stirred for 2 hours. After this time, solvent was removed under reduced pressure and the resultant solid suspended in 50 ml toluene. MAO (4.2 ml, 6.0mmol, 300 equivalents) was added and an immediately a green solution was observed. The solution placed under an ethylene atmosphere (1 bar). An immediate exotherm was observed. The reaction was run for 60 minutes during which time the vessel was left open to a supply of ethylene at 1 bar. The catalyst was then destroyed by addition of 50ml dilute aqueous HCl, the organic layer separated and dried over MgSO₄. The product mass, recorded by weighing the mass gain of the Schlenk reaction 25 vessel, was 10.3g.

GCMS analysis of the reaction products gave the following product distribution:

	<u>wt% Total Product</u>
Butenes	0.04
1-Hexene	82.17
2-Hexene	0.44
3-Hexene	0.15
1-Octene	1.37
Decenes	14.39
C12 olefins	0.20
C14 olefins	0.78
C16 olefins	0.00
C18 olefins	0.00

EXAMPLE 2

5 The procedure of Example 1 was followed, with the exception that 300 equivalents of MMAO (4.2 ml, 6.0 mmol) was used in place of MAO. The product mass was 8.8g.

EXAMPLE 3

10 The procedure of Example 1 was followed, with the exception that 100 equivalents of (*i*Bu₂AlO)₂ (2.0M solution in toluene, 1.0 ml, 2.0 mmol) was used in place of MAO. The product mass was 1.3 g.

EXAMPLE 4

The procedure as Example 1 was followed with the exception that CrCl₂ (3 mg, 0.02 mmol) was used in place of CrCl₃(THF)₃. The product mass was 5.6g.

15 EXAMPLE 5

A Schlenk vessel was charged with Cr(p-tolyl)Cl₂(THF)₃ (9 mg, 0.02 mmol) and (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ (10 mg, 0.02 mmol), 50 ml toluene was added, and the solution stirred for 5 minutes. MMAO (4.2 ml, 6.0 mmol) 300 equivalents) was added and the solution placed under an ethylene atmosphere (1 bar).

20 The reaction was run for 60 minutes during which time the vessel was left open to a supply of ethylene at 1 bar. The reaction was worked-up as described in example 1. The

product mass was 11.0g.

EXAMPLE 6

The procedure as Example 1 was followed with the exception that 0.04 mmol (20 mg) of (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ was used rather than 0.02 mmol. The product mass was 9.5g.

EXAMPLE 7

The procedure as Example 2 was followed with the exception that 0.01 mmol (5 mg) of (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ was used rather than 0.02 mmol. The product mass was 3.3g.

EXAMPLE 8

A Schlenk tube was charged with (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ (415mg, 0.8mmol) and CrCl₃(THF)₃ (300mg, 0.8 mmol) and 30ml dichloromethane added. A bright blue solution formed almost immediately which was stirred for 2 hours. After this time solvent was removed under reduced pressure to yield a blue solid; this was washed with diethyl ether and dried in vacuo. A further Schlenk tube was charged with mg of this compound and 50 ml toluene added. MMAO (16.8 ml, 24 mmol, 300 equivalents) was added and the solution placed under an ethylene atmosphere (1 bar). The reaction was run for 60 minutes during which time the vessel was left open to a supply of ethylene at 1 bar. The reaction was worked-up as described in example 1. The product mass was 2.5g.

EXAMPLE 9

Preparation of MAO on silica

Toluene (200 ml) was added to a vessel containing silica (prepared according to procedures described in WO 99/12981 example 37.1. Silica was supplied by Crosfield as grade ES70X), calcined at 200°C overnight, 20.5g after calcination) under an inert atmosphere. The slurry was mechanically stirred and MAO (1.5 M, 62.1 mmol, 41.4 ml) was added *via* syringe. The mixture was stirred for 1 hour at 80°C before removing excess toluene and drying under vacuum to obtain 15% w/w MAO on silica in quantitative yield.

30 Trimerisation using a supported catalyst composition

A Schlenk vessel was charged with CrCl₃(THF)₃ (8 mg, 0.02 mmol) and (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ (10mg, 0.02 mmol), 10ml THF was

added and the solution stirred for 2 hours. After this time, solvent was removed under reduced pressure and the resultant solid suspended in 20 ml toluene. MAO (1.4 ml, 2mmol, 100 equivalents) was added and an immediately a green solution was observed. This solution was then transferred via cannula to a Schlenk tube containing a slurry of 5 15% w/w MAO on silica (prepared as described above) in toluene (1g of MAO/Silica in 30 ml toluene). The green colour of the solution was quickly transferred onto the silica/MAO and a colourless supernatant remained. This slurry was stirred and placed under an ethylene atmosphere (1 bar). The reaction was run for 60 minutes during which time the vessel was left open to a supply of ethylene at 1 bar. The reaction was 10 worked-up as described in example 1. The product mass was 8.9g.

	wt% Total Product
1-Hexene	62
Octenes	0.28
Decenes	30.3

EXAMPLE 10

A Schlenk vessel was charged with $\text{CrCl}_3(\text{THF})_3$ (8 mg, 0.02 mmol) and (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂ (10 mg, 0.02 mmol), 10 ml THF was 15 added and the solution stirred for 2 hours. After this time, solvent was removed under reduced pressure, the resultant solid suspended in 10 ml toluene and MAO (4.2 ml, 6.0 mmol, 300 equivalents) added. This solution was then injected into an autoclave at 8 bar ethylene pressure and 50°C. The diluent was isobutane. The reaction was run for 1 20 hour at 8 bar ethylene pressure and 50°C after which time ethylene and isobutane gases were vented. The reaction products were then worked up as described in Example 1. The mass of product recovered was 40.0g and the productivity over one hour was 2000 g/mmol.h. GCMS analysis gave the following product distribution:

	wt% Total Product
Butenes	0.00
1-Hexene	88.37
2-Hexene	0.12
3-Hexene	0.00
Octenes	3.95
Decenes	6.61
C12 olefins	0.33
C14 olefins	0.20
C16 olefins	0.00
C18 olefins	0.00

EXAMPLE 11

The procedure of Example 10 was followed with the following exceptions:

5 500ml toluene diluent was used in place of isobutene and 0.01mmol of catalyst was used. The reactor conditions were maintained at 50°C and 8 bar ethylene pressure over the 60 minute run time. A stable gas uptake profile over the run time was observed. The mass of product recovered was 72.7 g and the productivity over one hour was 7270 g/mmol.h (134 700 g/gCr.h.)

10

	wt% Total Product
1-Hexene	86
Octenes	1.8
Decenes	8.7

EXAMPLE 12

The procedure of Example 11 was followed with the exceptions that the reactor conditions were maintained at 80°C and 20 bar ethylene pressure over the 60 minute run time. 0.0025mmol of catalyst was used. The mass of product recovered was 141g and the productivity over one hour was 56400 g/mmol.h (1 033 200 g/gCr.h.)

	wt% Total Product
1-Hexene	88.8
Octenes	1.8
Decenes	7.4

EXAMPLE 13

The procedure of Example 11 was followed with the exceptions that the reactor conditions were maintained at 108°C and 8 bar ethylene pressure over the 60 minute run time. 0.01mmol of catalyst was used. The mass of product recovered was 51.6 g and the productivity over one hour was 5160 g/mmol.h (95 900g/gCr.h)

	wt% Total Product
1-Hexene	86.6
Decenes	11

EXAMPLE 14

10 The procedure of Example 11 was followed with the exceptions that 1 bar of hydrogen was added to the reactor before the run. 0.01mmol of catalyst was used. The mass of product recovered was 94.7 g and the productivity over one hour was 9470 g/mmol.h (175 300 g/gCr.h.)

	wt% Total Product
1-Hexene	82
Octenes	0.45
Decenes	13.2

15

EXAMPLE 15

The procedure of Example 11 was followed with the exception that 0.01mmol of a supported catalyst, prepared as described in Example 8, was used. The mass of product recovered was 49.8 g and the productivity over one hour was 4980 g/mmol.h (90 406 g/gCr.h.)

	wt% Total Product
1-Hexene	89
Octenes	0.58
Decenes	7.9

EXAMPLE 16

The procedure of Example 11 was followed with the exceptions that 100ml of 1-butene was added to the reactor before the run and 400ml of toluene diluent was used.

5 The reactor conditions were maintained at 80°C and 4 bar ethylene pressure. 0.02mmol of catalyst was used. The mass of product recovered was 49.4g and the productivity over one hour was 2470 g/mmol.h (46125 g/gCr.h.)

	wt% Total Product
1-Hexene	60
Octenes	25
Decenes	10.9

EXAMPLE 17

The procedure of Example 1 was followed with the exceptions that the run time in this case was 90 minutes and the product mass was recorded by weighing the mass gain of the Schlenk reaction vessel at various times through the run.

Time (mins)	15	30	45	60	90
Mass gain (g)	2.7	5.2	7.6	10.0	13.0

15 GCMS analysis of the product after 90 minutes gave the following product distribution:

Product	wt%	Total
Butenes	0.00	
1-Hexene	64.10	
2-Hexene	0.13	
3-Hexene	0.00	
Octenes	0.44	
Decenes	28.93	
C12 olefins	0.13	
C14 olefins	4.99	
C16 olefins	0.00	
C18 olefins	0.59	

EXAMPLE 18

The procedure of Example 2 was followed with the exceptions that 20ml of toluene was used and 20ml of 1-dodecene was added at the start of the run. The product mass was 2.1g.

	wt% Total Product
1-Hexene	37
Decene	27
C16 olefins	29

EXAMPLE 19

The procedure of Example 2 was followed with the exceptions that 20ml of toluene was used and 20ml of 1-tetradecene was added at the start of the run. The product mass was 3.2g.

	wt% Total Product
1-Hexene	35.3
Decene	6.7
C18 olefins	50.8

EXAMPLE 20

The procedure of Example 9 was followed with the exceptions that 20ml of toluene was used and 20ml of 1-dodecene was added at the start of the run, in this case the run was for 4.5 hours. The product mass was 7.5g.

	wt% Total Product
1-Hexene	38
Decene	24
C16 olefins	38

5 EXAMPLE A (comparative)

The procedure of Example 1 was followed with the exceptions that 1,2-bis(diphenylphosphino)ethane (8mg, 0.02 mmol) was used in place of (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂. No product was formed.

EXAMPLE 21

10 The procedure of Example 1 was followed, with the exception that tris(2-methoxymethoxyphenyl)phosphane (18mg, 0.04 mmol) was used in place of (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂. The product mass was 1.2g.

GCMS analysis of the reaction products gave the following product distribution:

	wt% Total Product
1-Hexene	90.66
2-Hexene	2.94
1-Octene	2.85
Decenes	3.54

15 EXAMPLE B (comparative)

The procedure of Example 20 was followed with the exception that triphenylphosphane (11mg, 0.04 mmol) was used in place of tris(2-methoxymethoxyphenyl)phosphane. No product was formed

EXAMPLE 22

20 (Co)polymerisation of Ethylene

An autoclave was charged with isobutane (500ml) and triethylaluminium (2.0M solution in toluene, 1.5ml, 3mmol). The autoclave was pressurized to 8 bar ethylene pressure and heated to 50°C.

A catalyst (0.02mmol), prepared as described in Example 8, was then injected as 5 a slurry in 10 ml toluene. Almost immediately, a slurry of Ziegler catalyst (0.05g), prepared according to US 5470812, example A, was injected as a slurry in 10 ml toluene. The reaction was run for 1 hour at 8 bar ethylene pressure and 50°C after which time ethylene and isobutane gases were vented. The resultant polymer was washed with dilute aqueous HCl and then methanol and dried in vacuo. The mass of polymer 10 recovered was 36.0 g. NMR spectroscopy of the polymer shows the presence of butyl branches, indicating that an ethylene/1-hexene copolymer was produced.

EXAMPLE 23

(Co)polymerisation of Ethylene

A supported catalyst (0.01mmol) was prepared as described in Example 9 in 40 ml toluene. In a separate Schlenk tube, [rac-(ethylene bridged bis indenyl) zirconium dichloride] (mg, 0.01mmol) was dissolved in 10 ml toluene and MMAO (7ml, 10.0 mmol, 1000 equivalents) added. This second solution was added via canula to the supported catalyst slurry and the resultant slurry stirred under an ethylene atmosphere at 1 bar. The reaction was run for 60 minutes during which time the vessel was left open 20 to a supply of ethylene at 1 bar. The catalysts were then destroyed by careful addition of 50 ml dilute aqueous HCl. Both organic and aqueous fractions were then added to 500 ml of acetone, causing precipitation of the polymer produced. The polymer was washed with further portions of acetone and dried in vacuo. The mass of polymer recovered was 3.4g. NMR spectroscopy of the polymer shows the presence of butyl branches, 25 indicating that an ethylene/1-hexene copolymer was produced.

CLAIMS

1. Catalyst comprising
 - (a) a source of chromium, molybdenum or tungsten;
 - (b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but 5 excluding the case where all such polar substituents are phosphane, arsane or stibane groups; and optionally
 - (c) an activator.
2. Catalyst according to claim 1, wherein the catalyst is supported.
3. Supported catalyst having a productivity per mole of catalyst of at least 50%, 10 preferably at least 70% of its productivity when unsupported, which catalyst preferably comprises
 - (a) a source of a Group 3 to 10 transition metal;
 - (b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but 15 excluding the case where all such polar substituents are phosphane, arsane or stibane groups; and optionally
 - (c) an activator.
4. Catalyst according to claim 3, wherein the productivity is for trimerisation of olefins.
- 20 5. Catalyst according to claim 3 or 4, wherein the support is selected from silica, alumina, MgCl₂, zirconia, polyethylene, polypropylene, polystyrene, or poly(aminostyrene).

6. Catalyst according to claim 3, 4 or 5, wherein component (a) is a source of chromium, molybdenum or tungsten.

7. Catalyst according to any preceding claim, wherein component (a) is a source of chromium.

5 8. Catalyst according to any preceding claim, wherein the ligand of component (b) has the formula

$(R^1)(R^2)X-Y-X(R^3)(R^4)$ or

$X(R^1)(R^2)(R^3)$, wherein

X is phosphorus, arsenic or antimony;

10 Y is a linking group;

and R^1 , R^2 , R^3 and R^4 are each independently hydrocarbyl, substituted hydrocarbyl, heterohydrocarbyl or substituted heterohydrocarbyl groups, at least one of which in each formula has a polar substituent which is not a phosphane, arsane or stibane group, and any of the groups R^1-R^4 may independently be linked to one or more of each other or to 15 the bridging group Y, to form a cyclic structure together with X or X and Y.

9. Catalyst according to any preceding claim, wherein X is phosphorus.

10. Catalyst according to claim 8 or 9, wherein the optionally substituted hydrocarbyl or heterohydrocarbyl groups of R^1 , R^2 , R^3 and R^4 are independently selected from methyl, ethyl, ethylenyl, propyl, butyl, cyclohexyl, benzyl, phenyl, tolyl, xylyl, mesityl, 20 biphenyl, naphthyl, anthracenyl, methoxy, ethoxy, phenoxy (i.e. $-OC_6H_5$), tolyloxy (i.e. $-OC_6H_4(CH_3)$), xylyloxy, mesityloxy, dimethylamino, diethylamino, methylethylamino, thiomethyl, thiophenyl, trimethylsilyl or dimethylhydrazyl.

11. Catalyst according to any of claims 8 to 10, wherein those of R^1 to R^4 having polar substituents are each independently substituted phenyl, substituted naphthyl or 25 substituted anthracenyl groups.

12. Catalyst according to claim 11, wherein the polar substituents are independently selected from methoxy, ethoxy, isopropoxy, C_3-C_{20} alkoxy, phenoxy, pentafluorophenoxy, trimethylsiloxy, dimethylamino, methylsulphanyl, tosyl, methoxymethyl, methylthiomethyl, 1,3-oxazolyl, methoxymethoxy, hydroxyl, amino, 30 sulphate, nitro, phosphane, arsane and stibane.

13. Catalyst according to claim 12, wherein those of R^1 to R^4 having polar substituents are each independently o-methoxy phenyl or o-methoxymethoxy phenyl.

14. Catalyst according to any one of claims 8 to 13, wherein all of R¹ to R⁴ independently have a polar substituent which is not a phosphane, arsane or stibane group.

15. Catalyst according to any one of claims 8 to 14, wherein Y is a hydrocarbyl, substituted hydrocarbyl, heterohydrocarbyl, substituted hydrocarbyl or substituted heterohydrocarbyl bridging group, or an inorganic bridging group.

16. Catalyst according to claim 15, wherein the Y is methylene, 1,2-ethane, 1,2-phenylene, 1,3-propane, 1,2-catechol, 1,2-dimethylhydrazine, or -N(R⁵)- where R⁵ is hydrogen, hydrocarbyl or substituted hydrocarbyl.

17. Catalyst according to claim 16, wherein Y is -N(R⁵)-, and R⁵ is hydrogen, C₁-C₆ alkyl or phenyl.

18. Catalyst according to any preceding claim, wherein component (b) is selected from the following:

(2-methoxyphenyl)(phenyl)PN(Me)P(phenyl)₂

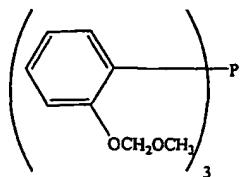
15 (2-methoxyphenyl)₂PN(Me)P(phenyl)₂
 (2-methoxyphenyl)(phenyl)PN(Me)P(2-methoxyphenyl)(phenyl)
 (2-methoxyphenyl)₂PN(Me)P(2-methoxyphenyl)₂
 (2-ethoxyphenyl)₂PN(Me)P(2-ethoxyphenyl)₂
 (2-isopropoxyphenyl)₂PN(Me)P(2-isopropoxyphenyl)₂

20 (2-hydroxyphenyl)₂PN(Me)P(2-hydroxyphenyl)₂
 (2-nitrophenyl)₂PN(Me)P(2-nitrophenyl)₂
 (2,3-dimethoxyphenyl)₂PN(Me)P(2,3-dimethoxyphenyl)₂
 (2,4-dimethoxyphenyl)₂PN(Me)P(2,4-dimethoxyphenyl)₂
 (2,6-dimethoxyphenyl)₂PN(Me)P(2,6-dimethoxyphenyl)₂

25 (2,4,6-trimethoxyphenyl)₂PN(Me)P(2,4,6-trimethoxyphenyl)₂
 (2-dimethoxyphenyl)(2-methylphenyl)PN(Me)P(2-methylphenyl)₂
 [2-(dimethylamino)phenyl]₂PN(Me)P[2-(dimethylamino)phenyl]₂
 (2-methoxymethoxyphenyl)₂PN(Me)P(2-methoxymethoxyphenyl)₂
 (2-methoxyphenyl)₂PN(Ethyl)P(2-methoxyphenyl)₂

30 (2-methoxyphenyl)₂PN(Phenyl)P(2-methoxyphenyl)₂
 (2-methoxyphenyl)₂PN(Me)N(Me)P(2-methoxyphenyl)₂
 (2-methoxyphenyl)₂PCH₂P(2-methoxyphenyl)₂

(2-methoxyphenyl)₂PCH₂CH₂P(2-methoxyphenyl)₂
tri(2-methoxymethoxyphenyl)phosphane i.e.



tri(2-methoxyphenyl) phosphane.

- 5 19. Catalyst according to any preceding claim, wherein component (c) is selected from trimethylaluminium (TMA), triethylaluminium (TEA), tri-isobutylaluminium (TIBA), tri-n-octylaluminium, methylaluminium dichloride, ethylaluminium dichloride, dimethylaluminium chloride, diethylaluminium chloride, ethylaluminiumsesquichloride, methylaluminiumsesquichloride, alumoxanes, tetrafluoroboric acid etherate, silver 10 tetrafluoroborate, sodium hexafluoroantimonate, boroxines, NaBH₄, trimethylboron, triethylboron, dimethylphenylammoniumtetra(phenyl)borate, trityltetra(phenyl)borate, triphenylboron, dimethylphenylammonium tetra(pentafluorophenyl)borate, sodium tetrakis[(bis-3,5-trifluoromethyl)phenyl]borate, H⁺(OEt₂)₂[(bis-3,5-trifluoromethyl)phenyl]borate, trityltetra(pentafluorophenyl)borate and 15 tris(pentafluorophenyl) boron, or mixtures thereof.
- 20. Catalyst for the trimerisation and polymerisation of 1-olefins, further comprising one or more catalysts (d) suitable for the polymerisation, oligomerisation or other chemical transformation of olefins.
- 21. Catalyst according to claim 20, wherein catalyst (d) is selected from Ziegler-Natta catalysts, metallocene catalysts, monocyclopentadienyl or "constrained geometry" catalysts, heat activated supported chromium oxide catalysts, late transition metal catalysts, and single site polymerisation catalysts.
- 22. Catalyst according to any preceding claim, which has a trimerisation productivity of at least 15000g product per mmol catalyst per hour, preferably at least 30000g 25 product per mmol catalyst per hour, at a temperature of 110°C or less and an ethylene partial pressure of 21 bar or less.
- 23. Catalyst for the trimerisation of olefins, wherein the catalyst productivity decays at a rate of less than 10 % per hour.
- 24. Catalyst for the trimerisation of ethylene, which has a trimerisation productivity of 30 at least 15000g product per mmol catalyst per hour, preferably at least 30000g product

per mmol catalyst per hour, at a temperature of 110°C or less and an ethylene partial pressure of 21 bar or less.

25. Catalyst according to any of claims 22 to 24 which is supported, preferably on a support selected from silica, alumina, MgCl₂, zirconia, polyethylene, polypropylene, 5 polystyrene, or poly(aminostyrene).

26. Process for the trimerisation of olefins, comprising contacting a monomeric olefin or mixture of olefins under trimerisation conditions with a catalyst which comprises

(a) a source of a Group 3 to 10 transition metal;

(b) a ligand containing at least one phosphorus, arsenic or antimony atom bound to at 10 least one hydrocarbyl or heterohydrocarbyl group having a polar substituent, but excluding the case where all such polar substituents are phosphane, arsane or stibane groups; and optionally

(c) an activator.

27. Process according to claim 26, wherein the catalyst is as defined in any of claims 1 15 to 25.

28. Process according to claim 26 or 27, wherein the olefin or mixture of olefins is additionally contacted with a further catalyst (d) suitable for the polymerisation, oligomerisation or other chemical transformation of olefins, such that the trimerisation products are incorporated into a higher polymer or other chemical product.

29. Process according to any one of claims 26 to 28, wherein the monomeric olefin is 20 ethylene.

30. Process according to any one of claims 26 to 28, wherein the mixture of olefins comprises ethylene and one or more C₃-C₃₆ monoolefin.

31. Process according to claim 30, wherein the C₃-C₃₆ monoolefin is a C₄-C₂₀ 25 monoolefin.

32. Process according to claim 31, wherein the C₄-C₂₀ monoolefin comprises butene, hexene, decene, a C₁₂ olefin or a C₁₄ olefin.

33. Process according to any one of claims 26 to 32, wherein at least 85 wt%, preferably at least 90 wt% of the trimerisation reaction product is one of the following:

30 1-hexene, 1-octene, 1-decene, a C₁₂ olefin, a C₁₄ olefin, a C₁₆ olefin or a C₁₈ olefin.

34. Process according to any one of claims 26 to 33, wherein the reaction temperature is less than 100°C, and/or the reaction pressure is below 30 bara.

35. Process according to any one of claims 26 to 34, wherein the residence time in the polymerisation reactor is less than 4 hours, preferably less than 3 hours.
36. Process according to any one of claims 26 to 35 wherein the reaction conditions are solution phase, slurry phase or gas phase.
- 5 37. Process according to claim 36 wherein the reaction is conducted under gas phase fluidised bed conditions.
38. Process according to claim 28, wherein the trimerisation reactor is upstream or downstream of at least one polymerisation or oligomerisation reactor.
- 10 39. Process according to claim 28, wherein the trimerisation reactor is incorporated into the reaction loop of at least one polymerisation or oligomerisation reactor.
40. Process according to claim 39, wherein the trimerisation reactor is incorporated into a side-stream taken from said reaction loop.
41. Process according to claim 28, wherein the trimerisation reaction product is produced in or introduced into at least one polymerisation or oligomerisation reactor.
- 15 42. Process according to any one of claims 28 and 38 to 40, wherein at least one trimerisation reaction product is separated from the remainder of the trimerisation reaction products prior to (re)introduction into the polymerisation or oligomerisation reactor.
43. Process according to any one of claims 26 to 42, wherein the trimerisation reaction is conducted in the presence of hydrogen and/or a halide source.
- 20 44. Use of a mixture of components (a), (b), optionally (c) and optionally (d), as defined in any of claims 1 to 25, as a catalyst for the trimerisation of olefins.
45. Use of a mixture of components (a), (b), (d) and optionally (c), as defined in any of claims 1 to 25, as a catalyst for the homopolymerisation of ethylene to produce polyethylene having a density of 960g/cm³ or less, preferably 940g/cm³ or less and more preferably 920g/cm³ or less.

INTERNATIONAL SEARCH REPORT

Int'l Application No
PCT/GB 01/03006

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7	B01J31/18	B01J31/34	B01J31/24	C07F9/50	C07F9/46
C07C2/36	C07C11/107		C08F10/02		

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 B01J C07F C07C C08F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, INSPEC, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 689 437 A (MURRAY REX E) 25 August 1987 (1987-08-25) column 18; claims; examples 37-40,43 see structures 15, 20, 21 column 7, line 1 - line 46 ----- US 5 968 866 A (WU FENG-JUNG) 19 October 1999 (1999-10-19) cited in the application the whole document -----	1,7-12, 19,26, 27, 29-32, 34-36, 43,44
A	US 5 968 866 A (WU FENG-JUNG) 19 October 1999 (1999-10-19) cited in the application the whole document -----	1,7-19, 26,27, 29-32, 34-36, 43,44

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the International filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the International filing date but later than the priority date claimed

- *T* later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the International search

4 October 2001

Date of mailing of the international search report

23/10/2001

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
 NL - 2280 HV Rijswijk
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
 Fax: (+31-70) 340-3016

Authorized officer

Goebel, M

INTERNATIONAL SEARCH REPORT

Inte	nal Application No
PCT/GB 01/03006	

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 4 472 525 A (SINGLETON DAVID M) 18 September 1984 (1984-09-18) claims; examples ---	1-19, 26, 27, 29-32, 34-36, 43, 44
P, A	WO 01 10876 A (WASS DUNCAN FRANK ;BP CHEM INT LTD (GB)) 15 February 2001 (2001-02-15) page 4, line 16; claims 12-15, 19-22; examples 4, 28 page 11, line 10 - line 16 ---	1-21, 26-32, 34-45
A	WO 97 37765 A (BP CHEM INT LTD ;DOSSETT STEPHEN JOHN (GB)) 16 October 1997 (1997-10-16) cited in the application page 8, line 30 - line 32; claims; example 12 ---	1-19
A	US 5 668 249 A (BAARDMAN FRANK ET AL) 16 September 1997 (1997-09-16) the whole document ---	1-19
A	US 5 856 610 A (HINO TAKAHIRO ET AL) 5 January 1999 (1999-01-05) column 2, line 65 -column 3, line 11; claims; figures; example 1 ---	20, 21, 28-32, 34-45
A	US 3 300 458 A (WALKER WELLINGTON E ET AL) 24 January 1967 (1967-01-24) column 6, line 66 -column 7, line 29; example 1; table 1 ---	20, 21, 28-32, 45
X	DATABASE CA 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; ARIFFIN, AZHAR ET AL: "The asymmetric synthesis of phosphorus- and sulfur-containing tricarbonyl(.eta.6-arene) chromium complexes using the chiral base approach" retrieved from STN Database accession no. 132:108059 XP002179252 see structure abstract & J. CHEM. SOC., PERKIN TRANS. 1 (1999), (21), 3177-3189 , -- -/-	1, 7-13

INTERNATIONAL SEARCH REPORT

Inte	nal Application No
PCT/GB 01/03006	

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>DAHLENBURG L ET AL: "Coordination chemistry of functional phosphines. VIII. Tetracarbonyl complexes of tungsten and molybdenum with 2-(diphenylphosphinyl)aniline ligands" JOURNAL OF ORGANOMETALLIC CHEMISTRY, ELSEVIER-SEQUOIA S.A. LAUSANNE, CH, vol. 585, no. 2, 15 August 1999 (1999-08-15), pages 225-233, XP004177462 ISSN: 0022-328X the whole document</p> <p>---</p>	1,7-13, 19
X	<p>PIETSCH J ET AL: "Coordination chemistry of functional phosphines II.. Carbonyl(nitrosyl) tungsten complexes with 2-diphenylphosphinoanisole and 2-diphenylphosphinoaniline, -benzoate and -phenolate as ligands" JOURNAL OF ORGANOMETALLIC CHEMISTRY, ELSEVIER-SEQUOIA S.A. LAUSANNE, CH, vol. 495, no. 1, 14 June 1995 (1995-06-14), pages 113-125, XP004023911 ISSN: 0022-328X the whole document</p> <p>---</p>	1,7-13, 19
X	<p>DATABASE CA 'Online' CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US;</p> <p>BONI, GILLES ET AL: "Heterobimetallic Dibridged Complexes 'Cp₂Ta(.mu.-CO)(.mu.-PM₂)M'(CO)4! (M' = Cr, W): Synthesis and Reactivity Toward Two-Electron Donor Ligands L (L = PR₃, Me₂P(CH₂)_nPM₂, CNR)" retrieved from STN Database accession no. 124:117494 XP002179253 see structure abstract & ORGANOMETALLICS (1995), 14(12), 5652-6 ,</p> <p>---</p> <p>-/-</p>	1,7-13

INTERNATIONAL SEARCH REPORT

Inte	nal Application No
PCT/GB 01/03006	

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>DATABASE CA 'Online!' CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; DUNBAR, KIM R. ET AL: "Carbon Monoxide Reactions of the Fluxional Phosphine Complex (.eta.3-PR₃) Mo (CO)₃ (R = 2,4,6-Trimethoxyphenyl)" retrieved from STN Database accession no. 121:72401 XP002179254 see structure abstract & ORGANOMETALLICS (1994), 13(7), 2713-20 ,</p> <p>---</p>	1,7-14
X	<p>DATABASE CA 'Online!' CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; DUNBAR, K. R. ET AL: "Structure of 'HTMPP!2W2C19 'HTMPP = tris(2,4,6- trimethoxyphenyl)phosphonium!" retrieved from STN Database accession no. 114:175345 XP002179255 see structures abstract & ACTA CRYSTALLOGR., SECT. C: CRYST. STRUCT. COMMUN. (1991), C47(1), 23-6 ,</p> <p>---</p>	1,7-14
X	<p>DATABASE CA 'Online!' CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; BURGESS, KEVIN ET AL: "Stereochemically matched (and mismatched) bisphosphine ligands: DIOP-DIPAMP hybrids" retrieved from STN Database accession no. 117:212586 XP002179256 see structures abstract & ORGANOMETALLICS (1992), 11(11), 3588-600</p> <p>,</p> <p>---</p> <p>-/-</p>	1,7-16

INTERNATIONAL SEARCH REPORT

Inte	nal Application No
PCT/GB 01/03006	

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>HIRSIVAARA L ET AL: "Organometallic derivatives of multidentate phosphines 'o-(methylthio)phenyl!diphenylphosphine and bis(o-(methylthio)phenyl(phenylphosphine: preparation and characterization of group 6 metal carbonyl derivatives"</p> <p>JOURNAL OF ORGANOMETALLIC CHEMISTRY, ELSEVIER-SEQUOIA S.A. LAUSANNE, CH, vol. 579, no. 1-2, 5 May 1999 (1999-05-05), pages 45-52, XP004170546</p> <p>ISSN: 0022-328X the whole document</p> <p>---</p>	1,7-13
P,X	<p>DATABASE CA 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US;</p> <p>HIRSIVAARA, LEENI ET AL: "M(CO)₆ (M = Cr, Mo, W) derivatives of (o-anisyl)diphenylphosphine, bis(o-anisyl)phenylphosphine tris(o-anisyl)phosphine and (p-anisyl)bis(o-anisyl)phosphine" retrieved from STN Database accession no. 134:36335 XP002179257 see structures abstract & INORG. CHIM. ACTA (2000), 307(1-2), 47-56 ,</p> <p>---</p>	1,7-14, 18

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Claims Nos.: 22-25 33

1. Present claims 22-25, 33 relate to a compound (catalyst) and method (catalytic olefin trimerisation) defined (solely) by reference to several desirable characteristics or properties, namely (a) a stated specific productivity (dependent catalyst claim 22 and independent catalyst claim 24), (b) a stated specific catalyst productivity decay (independent catalyst claim 23), (c) a specified selectivity for a list of products obtainable by the method.

The claims cover all compounds/methods having this (further) characteristic or property, whereas the application provides support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT for only a very limited number of such compounds/methods. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search is impossible. Independent of the above reasoning, the claims also lack clarity (Article 6 PCT). An attempt is made to define the compound/method by reference to a result to be achieved. Again, this lack of clarity in the present case is such as to render a meaningful search impossible. Consequently, no search has been carried out for these claims.

2. Present claims 1-8, 14, 15, 20, 26-28, 44 and 45 relate to an extremely large number of possible compounds/methods, containing so many options, variables and possible permutations that a lack of clarity (and/or conciseness) within the meaning of Article 6 PCT arises to such an extent as to render a meaningful search of the claims impossible. Additionally, the catalysts (used in the methods) are claimed in a highly inconsistent manner (claims 1, 3, 20, 26, 27, as well as claims 23, 24 stated under item 1 above, having not a single mandatory feature in common), their definition relies on further desiderata (claims 3, 4), relative terms ("polar substituent", claims 1, 3, 8, 14, 26) and vague terms ("suitable for other transformations", claims 20/28), while the use in claim 45 contradicts the description (copolymerisation exclusively described therein).

Furthermore, support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT is to be found only for a very small proportion of the compound/methods claimed. In the present case, the claims so lack clarity (and/or conciseness), support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Consequently, the search has been carried out for those parts of the claims which appear to be clear (and/or concise), supported and disclosed, namely those parts relating to the compounds/methods as follows (only search restrictions in claim n stated, giving searched claim n'; not mentioned dependent claims searched according to the restrictions defined in the respective independent and/or referred to claims; see also not searched claims in item 1):

2a. compound claims:

claim 1': ligands (b) as defined in claim 8' (see below).

claim 3': productivity as defined in claim 4; metals (a) = Cr, Mo, W (as in claim 1); ligands (b) as in claim 8' (see below); both components (a)

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

and (b) are mandatory.

claim 8': Y as defined in claim 15' (see below); R1-R4 as defined in claim 10 and close analogues thereof; polar substituents as defined in claim 12 and close analogues thereof.

claim 15': Y as defined in claim 16 and close analogues thereof.

claim 20': catalyst with components (a), (b) and optionally (c) must be present, as defined in claims 1' or 3' (see above) or thereupon dependent claims 2, 4-7, 8' (see above), 9-14, 15' (see above), 16-19; catalyst (d) suitable for either polymerisation or oligomerisation of olefins.

2b. method claims:

claim 26': catalyst with metals (a) = Cr, Mo, W (as in claim 1), ligands (b) as defined in claim 8' (see above).

claim 27': catalyst as defined in claims 1' or 3' (see above) or thereupon dependent claims 2, 4-7, 8' (see above), 9-14, 15' (see above), 16-19, 20' (see above), 21.

claim 28': catalyst (d) as in claim 20' (see above).

claim 44': catalyst as defined in claim 27' (see above).

claim 45': catalyst as defined in claim 27' (see above); copolymerisation of ethylene, to produce polymers with density of 960 g/dm³ or less.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

INTERNATIONAL SEARCH REPORT

Information on patent family members

Inte PC	al Application No 01/03006
------------	-------------------------------

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 4689437	A	25-08-1987	AT 86957 T AU 597397 B2 AU 7594987 A BR 8703795 A CA 1329608 A1 CN 1051312 A ,B DE 3784800 T2 EP 0254277 A2 ES 2046185 T3 FI 873199. A ,B, GR 3008105 T3 JP 2042250 C JP 7064759 B JP 63113006 A JP 2930921 B2 JP 10195000 A JP 2920126 B2 JP 10195089 A JP 2948181 B2 JP 10195090 A JP 2915302 B2 JP 7165821 A KR 9203609 B1 NO 173600 C NO 912815 A NZ 221148 A US 4822915 A US 4716138 A ZA 8705353 A	15-04-1993 31-05-1990 28-01-1988 29-03-1988 17-05-1994 15-05-1991 19-08-1993 27-01-1988 01-02-1994 22-01-1988 30-09-1993 09-04-1996 12-07-1995 18-05-1988 09-08-1999 28-07-1998 19-07-1999 28-07-1998 13-09-1999 28-07-1998 05-07-1999 27-06-1995 04-05-1992 05-01-1994 22-01-1988 26-04-1989 18-04-1989 29-12-1987 27-04-1988
US 5968866	A	19-10-1999	US 5811618 A US 5550305 A US 5744677 A CA 2115639 A1 DE 69410876 D1 DE 69410876 T2 EP 0622347 A1 JP 6298673 A CA 2079399 A1 DE 69212761 D1 DE 69212761 T2 EP 0537609 A2 JP 5221882 A	22-09-1998 27-08-1996 28-04-1998 04-09-1994 16-07-1998 08-10-1998 02-11-1994 25-10-1994 17-04-1993 19-09-1996 02-01-1997 21-04-1993 31-08-1993
US 4472525	A	18-09-1984	AU 570685 B2 AU 2900284 A CA 1219272 A1 DE 3463757 D1 EP 0128596 A2 JP 1739052 C JP 4025936 B JP 60011433 A NZ 208378 A US 4503280 A	24-03-1988 13-12-1984 17-03-1987 25-06-1987 19-12-1984 26-02-1993 06-05-1992 21-01-1985 29-05-1987 05-03-1985
WO 0110876	A	15-02-2001	AU 6000100 A WO 0110876 A1	05-03-2001 15-02-2001

INTERNATIONAL SEARCH REPORT

Information on patent family members

Intelli	nter	al Application No
PCT/GB 01/03006		

Patent document cited in search report		Publication date		Patent family member(s)		Publication date
WO 9737765	A	16-10-1997		AU 2302797 A EP 0850105 A1 WO 9737765 A1 JP 11506981 T US 6140460 A US 6018016 A		29-10-1997 01-07-1998 16-10-1997 22-06-1999 31-10-2000 25-01-2000
US 5668249	A	16-09-1997		CA 2170945 A1 EP 0733662 A1 JP 8283405 A		23-09-1996 25-09-1996 29-10-1996
US 5856610	A	05-01-1999		JP 9143228 A GB 2307478 A ,B SG 73436 A1		03-06-1997 28-05-1997 20-06-2000
US 3300458	A	24-01-1967		NL 6702036 A BE 695375 A FR 1513813 A US 3231550 A		12-08-1968 11-09-1967 09-05-1968 25-01-1966